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Declarations under Rule 4.17:

- as to applicant's entitlement to apply for and he granted a patent (Rule 4.17(ii)) for the following designations AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES. F1, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS. JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG. MK, MN, MW, MX, MZ, NO, NZ, OM, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SI, TJ, TM, TN, TR, TT, TZ, UA, UG. UZ, VC, VN, YU, ZA, ZM, ZW, ARIPO patent (GH, GM, KE, US, MW, MZ, SD, SI, SZ, TZ, UG, ZM, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PT, RO, SE, SI, SK, TR), OAPI patent (BE, BJ, CE, CG, CL, CM, GA, GN, GQ, GW, ML, MR, NE, SN. TD. TG)
- of inventorship (Rule 4.17(iv)) for US only

with international search report

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

(57) Abstract: The present invention provides novel crystalline forms of ziprasidone hydrochloride monohydrate, processes for their preparation and pharmaceutical compositions containing them.





NOVEL CRYSTALLINE FORMS OF ZIPRASIDONE HYDROCHLORIDE

FIELD OF THE INVENTION

The present invention provides novel crystalline forms of ziprasidone hydrochloride monohydrate, processes for their preparation and pharmaceutical compositions containing them.

BACKGROUND OF THE INVENTION

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Ziprasidone of formula (1):

$$0 \longrightarrow 1$$

$$0 \longrightarrow 1$$

$$0 \longrightarrow 1$$

or 5-[2-[4-(1,2-Benzisothiazol-3-yl)-1-piperazinyl]ethyl]-6-chloro-1,3-dihydro-2H-indol-2-one and its salts are antipsychotic agents. Ziprasidone hydrochloride and related compounds and their therapeutic uses are disclosed in US 4,831,031.

The crystalline forms of ziprasidone mesylate were reported in WO 97/42190, WO 97/42191.

It has now been discovered that ziprasidone hydrochloride monohydrate can be prepared in three stable crystalline forms having good dissolution characteristics.

The object of the present invention is to provide stable novel crystalline forms of ziprasidone hydrochloride monohydrate, processes for preparing these forms and pharmaceutical compositions containing them.

DETAILED DESCRIPTION OF THE INVENTION

In accordance with the present invention, there is provided a novel crystalline form of ziprasidone hydrochloride monohydrate, designated as form I,

characterized by an x-ray powder diffraction spectrum having peaks expressed as 20 at about 10.9, 13.9, 15.9, 16.4, 17.5, 19.2, 20.6, 21.3, 21.9, 24.2, 24.7, 24.9, 25.7, 25.9 and 28.9 degrees. Figure 1 shows typical form I x-ray powder diffraction spectrum.

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In accordance with the present invention, a process is provided for preparation of ziprasidone hydrochloride monohydrate form I. Thus, a mixture of ziprasidone free base, hydrochloric acid and water is heated to about 45°C to 100°C; and ziprasidone hydrochloride monohydrate form I is isolated by filtration or centrifugation. Preferably, the mixture of ziprasidone free base, hydrochloric acid and water is heated to about 55°C to 65°C; and ziprasidone hydrochloride monohydrate form I is isolated by filtration or centrifugation.

In accordance with the present invention, there is provided a novel crystalline form of ziprasidone hydrochloride monohydrate, designated as form II, characterized by an x-ray powder diffraction spectrum having peaks expressed as 26 at about 10.9, 11.3, 18.1, 19.5, 21.9, 23.7, 24.4, 24.8 and 26.2 degrees. Figure 2 shows typical form II x-ray powder diffraction spectrum.

In accordance with the present invention, a process is provided for preparation of ziprasidone hydrochloride monohydrate form II. Thus, ziprasidone free base, an alcohol or a mixture of alcohols, dimethylformamide, a chlorinated solvent, hydrochloric acid and water are mixed to form a solution of ziprasidone hydrochloride; and the solvents are removed by the techniques such as vacuum drying, spray drying, freeze drying and lyophilization to form ziprasidone hydrochloride monohydrate form II. Water may be directly mixed or it may be mixed, for example, as an aqueous solution of hydrochloric acid. The alcohols are selected from the group consisting of methanol, ethanol, isopropyl alcohol, tert-butyl alcohol and n-butyl alcohol. The preferable alcohols are methanol and ethanol. The chlorinated solvents are selected from the group consisting of methylene dichloride, chloroform, carbon tetrachloride and ethylene dichloride. The preferable ester solvents are chloroform and methylene dichloride.

In accordance with the present invention, there is provided a novel crystalline form of ziprasidone hydrochloride monohydrate, designated as form III, characterized by an x-ray powder diffraction spectrum having peaks expressed as 20 at about 10.9, 14.8, 15.9, 18.1, 19.5, 21.8, 24.3, 24.9, 25.9 and 26.5 degrees. Figure 3 shows typical form III x-ray powder diffraction spectrum.

In accordance with the present invention, a process is provided for preparation of ziprasidone hydrochloride monohydrate form III. Thus ziprasidone free base, an ether solvent or a mixture of ether solvents, dimethylformamide, hydrochloric acid and water are mixed to form a solution of ziprasidone hydrochloride; and ziprasidone hydrochloride monohydrate form III is isolated from the solution. Water may be directly mixed or it may be mixed, for example, as an aqueous solution of hydrochloric acid. The ether solvents are selected from the group consisting of diethyl ether, dilsopropyl ether and tert-butyl methyl ether. The preferable ether solvent is diethyl ether.

Ziprasidone free base used in the above processes can be obtained from the previously known methods.

In accordance with the present invention, there is provided a pharmaceutical composition comprising a crystalline form of ziprasidone hydrochloride monohydrate and pharmaceutically acceptable carrier or diluent.

The crystalline form includes form I, form II or form III.

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BRIEF DESCRIPTION OF THE DRAWINGS

Figure 1 is a x-ray powder diffraction spectrum of ziprasidone hydrochloride monohydrate form I.

Figure 2 is a x-ray powder diffraction spectrum of ziprasidone hydrochloride monohydrate form II.

Figure 3 is a x-ray powder diffraction spectrum of ziprasidone hydrochloride monohydrate form III.

x-Ray powder diffraction spectrum was measured on a Siemens D5000 x-ray powder diffractometer having a copper- $K\alpha$ radiation.

The following examples further illustrate the present invention.

Example 1

Ziprasidone free base (10 gm), conc. hydrochloric acid (10 ml) and water (150 ml) are mixed and the reaction mass is heated to 60°C and stirred for 4 hours at 60°C to 65°C. The contents are cooled to 25°C, filtered, washed with water and dried to give 10 gm of ziprasidone hydrochloride monohydrate form I.

Example 2

Ziprasidone free base (2.5 gm), methanol (100 ml), dimethylformamide (100 ml), chloroform (25 ml) and conc. hydrochloric acid (1.5 ml) are mixed at 25°C. The contents are heated to 60°C and stirred for 10 minutes at 60°C to 65°C and the clear solution thus obtained is subjected to vacuum drying at 70°C for 40 hours to give ziprasidone hydrochloride monohydrate form II in near quantitative yield.

Example 3

Ziprasidone free base (3.0 gm), methanol (120 ml), dimethylformamide (100 ml), chloroform (30 ml) and conc. hydrochloric acid (1.5 ml) are mixed at 25°C. The contents are heated to 60°C and stirred for 10 minutes at 60°C to 65°C and the clear solution thus obtained is subjected to spray drying to give ziprasidone hydrochloride monohydrate form II.

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Example 4

Ziprasidone free base(5.0 gm) is added to diethyl ether (50 ml) and heated to reflux temperature. Then dimethylformamide (145 ml) is added and the contents are stirred for 2 hours under reflux. Then conc. hydrochloric acid (2.5 ml) and water (3 ml) are added, and the solution is cooled to 25°C. The separated crystals are filtered to give 3.5 gm of ziprasidone hydrochloride monohydrate form III.

We claim:

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- 1. A crystalline ziprasidone hydrochloride monohydrate form I, characterized by an x-ray powder diffraction spectrum having peaks expressed as 28 at about 10.9, 13.9, 15.9, 16.4, 17.5, 19.2, 20.6, 21.3, 21.9, 24.2, 24.7, 24.9, 25.7, 25.9 and 28.9 degrees.
- 2. A crystalline ziprasidone hydrochloride monohydrate form I, further characterized by an x-ray powder diffraction spectrum as in figure 1.
- 3. A process for preparation of ziprasidone hydrochloride monohydrate form I as defined in claim 1, which comprises the steps of:
- a) mixing ziprasidone free base, hydrochloric acid and water;
- b) heating to about 45°C to 100°C; and
- c) isolating ziprasidone hydrochloride monohydrate form! by filtration or centrifugation.
- A process according to claim 3, wherein the reaction mass is heated to about 60°C to 65°C in (b).
 - 5. A crystalline ziprasidone hydrochloride monohydrate form II, characterized by an x-ray powder diffraction spectrum having peaks expressed as 20 at about 10.9, 11.3, 18.1, 19.5, 21.9, 23.7, 24.4, 24.8 and 26.2 degrees.
- 6. A crystalline ziprasidone hydrochloride monohydrate form II as defined in claim 5, further characterized by an x-ray powder diffraction spectrum as in figure 2.
 - 7. A process for preparation of ziprasidone hydrochloride monohydrate form II as defined in claim 5, which comprises the steps of:
- 25 a) mixing ziprasidone free base, an alcohol or a mixture of alcohols, dimethylformamide, a chlorinated solvent, hydrochloric acid and water to form a ziprasidone hydrochloride solution; and
- b) removing the solvents from the solution;
 wherein the alcohol is selected from the group consisting of methanol, ethanol,
 isopropyl alcohol, tert-butyl alcohol and n-butyl alcohol; and the chlorinated
 solvent is selected from the group consisting of methylene dichloride,
 chloroform, carbon tetrachloride and ethylene dichloride.

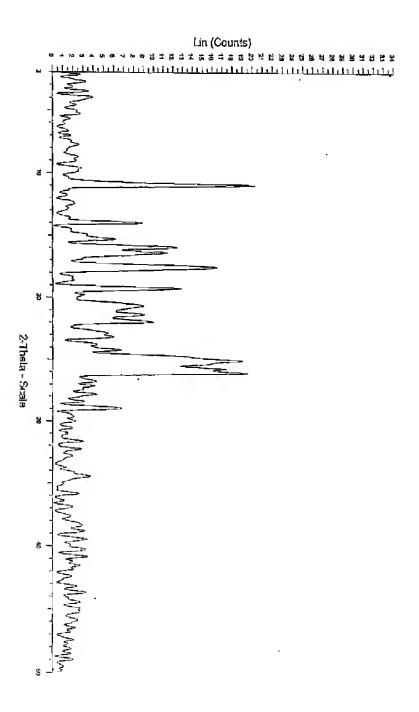
 A process according to claim 7, wherein the solvents are removed by the techniques such as vacuum drying, spray drying, freeze drying and lyophilization.

- 9. A process according to claim 7, wherein the alcohol is methanol.
- 10. A process according to claim 7, wherein the chlorinated solvent is chloroform.
 - 11. A crystalline ziprasidone hydrochloride monohydrate form III, characterized by an x-ray powder diffraction spectrum having peaks expressed as 20 at about 10.9, 14.8, 15.9, 18.1, 19.5, 21.8, 24.3, 24.9, 25.9 and 26.5 degrees.
- 10 12. A crystalline ziprasidone hydrochloride monohydrate form III as defined in claim 11, further characterized by an x-ray powder diffraction spectrum as in figure 3.
 - 13. A process for preparation of ziprasidone hydrochloride monohydrate form III as defined in claim 11, which comprises the steps of:
- a) mixing ziprasidone free base, an ether solvent or a mixture of ether solvents, dimethylformamide, hydrochloric acid and water to form a ziprasidone hydrochloride solution; and
 - b) isolating ziprasidone hydrochloride monohydrate form III from the solution; wherein the ether solvent is selected from the group consisting of diethyl ether, diisopropyl ether and tert-butyl methyl ether.

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- 14. A process according to claim 11, wherein the ether solvent is diethyl ether.
- 15. A pharmaceutical composition comprising ziprasidone hydrochloride monohydrate form I of claim 1 and a pharmaceutically acceptable carrier or diluent.
- 25 16. A pharmaceutical composition comprising ziprasidone hydrochloride monohydrate form II of claim 5 and a pharmaceutically acceptable carrier or diluent.
 - 17. A pharmaceutical composition comprising ziprasidone hydrochloride monohydrate form III of claim 11 and a pharmaceutically acceptable carrier or diluent.



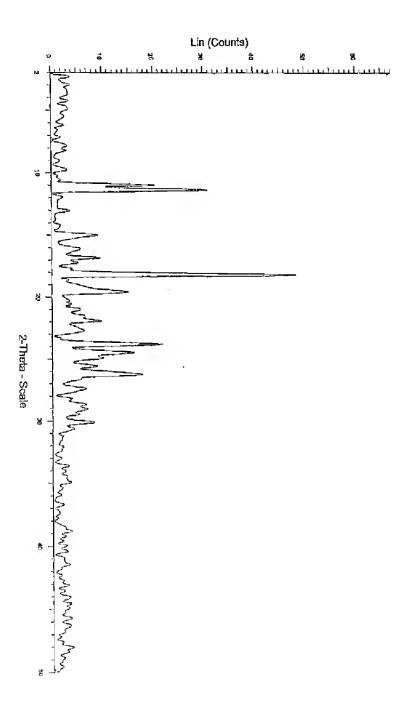
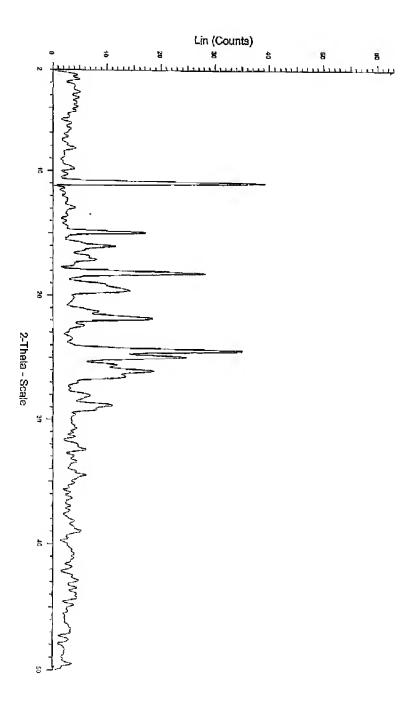


fig. 3



INTERNATIONAL SEARCH REPORT

International application No. PCT/IN 03/00154-0

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CLA	SSIFICATION OF SUBJECT MATTER	<u> </u>			
IPC ⁷ : C	07D 417/12; A61K 31/425				
According	to International Patent Classification (IPC) or to both na DS SEARCHED	tional classification and IPC			
Minimum	documentation searched (classification system followed	by classification symbols)			
_	07D, A61K				
	nition searched other than minimum documentation to the	extent that such documents are included in	n the fields searched		
Electronic	data base consulted during the international search (name	e of data base and, where practicable, sear	ch terms used)		
	UE; STN Karlsruhe/Registry, CAPLUS				
	CUMENTS CONSIDERED TO BE RELEVANT		d and		
Category	Citation of document, with indication, where appropriate	e, of the relevant passages	Relevant to claim No.		
x	EP 0586191 A1 (PFIZER INC.) 9 Ma abstract; page 2, line 23; page 2, lines page 2, line 50 - page 3, line 9; examp	: 31-32; page 2, lines 40-44;	1,3,4,15		
Α	, <u>2</u>		5,7-11,13,		
			14,16,17		
х	ED 0594009 A4 (DEIZER INC.), O.M.		4 7 4 45		
^	EP 0584903 A1 (PFIZER INC.) 2 Ma page 4, line 56 - page 5, line 8; claims		1,3,4,15		
Α	page 1, into co - page 5, inte 6, dants	5,7-11,13,			
•		14,16,17			
X	US 5338846 A (BUSCH et al.) 16 Au column 1, lines 40-47; column 2, lines 16; column 4, lines 3-14; claims 11,17	1,3,4,15			
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	her documents are listed in the continuation of Box C. 1 categories of cited documents:	See patent family annex. "I" later document published after the internal	1. 1.000		
"A" docum consid "E" carlier filing:	ent defining the general state of the art which is not cred to be of particular relevance application or patent but published on or after the international tate	date and not in conflict with the application the principle or theory underlying the inve "X" decement of particular relevance; the claim considered novel or cannot be considered.	n but cited to understand ntion ned invention cannot be		
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	n Patent Office er Straße 87, A-1200 Vienna	KOLLER G.			
	No. 1/53424/535	Telephone No. 1/53424/458			
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INTERNATIONAL SEARCH REPORT

International application No.
PCT/IN 03/00154-0

Box I	Observations where certain claims were found unsearchable (Continuation of item 1 of first sheet)						
This international search report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:							
I. 🗆	Claims Nos.: because they relate to subject matter not required to be scarched by this Authority, namely:						
2. 🔀	Claims Nos.: 2, 6, 12 because they relate to parts of the international application that do not comply with the prescribed requirements to such an extent that no meaningful international search can be carried out, specifically: Claims 2, 6, 12 refer to figures of the drawings and contravene Rule 6.2(a) PCT.						
3.	Claims Nos.: because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a).						
Box H	Observations where unity of invention is lacking (Continuation of item 2 of first sheet)						
). L	As all required additional search fees were timely paid by the applicant, this international search report covers all searchable claims.						
2.	As all searchable claims could be searched without effort justifying an additional fee, this Authority did not invite payment of any additional fee.						
4.	As only some of the required additional search fees were timely paid by the applicant, this international search report covers only those claims for which fees were paid, specifically claims Nos.: No required additional search fees were timely paid by the applicant. Consequently, this international search report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.:						
Remar	The additional search fees were accompanied by the applicant's protest. No protest accompanied the payment of additional search fees.						

INTERNATIONAL SEARCH REPORT

Information on patent family members

International application No. PCT/IN 03/00154-0

	Patent document cited in search report		Publication date	Patent family member(s)			Publication date
EP	A	584903	- JU	В	642836	1993-10-28	
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				sk	B	2805,848	2000-04-10
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				CZ	В	285984	1999-12-1
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				ES	T	2161703T	2001-12-1
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